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Physical Science EQT Study Guide Qtr. 1 Brewer 2 of 3 1.3 34. What would be the mass number of an atom of gold with 79 protons, 79 electrons, and 118 neutrons? 1.3 35. A particular isotope has 11 protons, 12 neutrons and 11 electrons. This is an isotope of which element? 1.3 36. A particular isotope has 6 protons, 8 neutrons and 6 electrons.

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