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Chapter 12 REVIEW: Stoichiometry,
Theoretical, Actual & Percent yield
Part I. Stoichiometry 1. $1 \text{ ZnI}_2 \rightarrow 1 \text{ Zn} + 1 \text{ I}_2$ How many grams of iodine will you produce if you begin your rxn with 56.7g of zinc iodide? 56.7 g 1 mole 1 mole 253.8 g 319.19 g 1 mole 1 mole = 45.08 grams of Iodine 2. $1 \text{ Pb}_2(\text{CO}_3)_3 + 6 \text{ K} \rightarrow 2 \text{ Pb} + 3 \text{ K}_2\text{CO}_3$

Chapter 12 REVIEW: Stoichiometry, Theoretical, Actual ...

STOICHIOMETRY 12 Vocabulary Review Match the correct vocabulary term to each numbered statement. Write the letter Of the correct term on the line. Column A 1. the starting materials in a chemical reaction 2. a conversion factor derived from the coefficients of a balanced chemical

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equation interpreted in terms Of moles

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Chemistry Chapter 12 vocabulary.
Stoichiometry. STUDY. PLAY. Mole.
 6.02×10^{23} representative particles.
Reactants. A starting material in a
chemical reaction. Products. The
elements or compounds produced by
a chemical reaction. Limiting reactant.

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Chapter 12 Quiz: Stoichiometry. You react 25.0 liters of oxygen gas at STP with 50.0 ml of ethanol liquid. Ethanol has a density of 0.789 g/ml. Use this reaction to solve the following problems.

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