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2.00pm (2nd Set) 1. A is

0.100 mol dm<sup>-3</sup> solution of

an acid. B is a solution KOH

containing 2.8 g per 500

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cm<sup>3</sup>. (a) Put A into the burette and titrate it against 25.0 cm<sup>3</sup> portions B using methyl orange as indicator. Repeat the titration to obtain consistent titres.

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09.30am - 11.30am (1st Set)  
Chemistry 3 (Practical)  
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2.00pm (2nd Set) 1. A is  
0.100 mol dm<sup>-3</sup> solution of

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an acid. B is a solution KOH containing 2.8 g per 500 cm<sup>3</sup>. (a) Put A into the burette and titrate it against 25.0 cm<sup>3</sup> portions B

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