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IB Chemistry HL & SL -

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– 2 – Section A Answer all questions.

Write your answers in the boxes provided.

1. Ethanedioic acid is a diprotic acid. A student determined the value of x in the formula of hydrated ethanedioic acid,

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$\text{HOOC-COOH} \cdot x\text{H}_2\text{O}$, by titrating a known mass of the acid with a $0.100 \text{ mol dm}^{-3}$ solution of $\text{NaOH}(\text{aq})$. 0.795 g of ethanedioic acid was dissolved in distilled water and made up to a total ...

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5. a ratio of electrons : aluminium ions 3 :
1 amount Al 13 1 1.296 10 C 96500Cmol
3 «» 4.48 107 «mol» mass Al « 4.48 107
mol 26.98 g mol 1» 1.21 109 «g» Award
[3] for correct final answer. 3 5.

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need to be learning during the school year in order to ace the IB Chemistry papers. Common errors students make when studying are: #1: Avoiding the topics you didn't fully comprehend in class.

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