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Created by. noTORiousChan. Terms for the first semester exam. Terms in this set (108) chemistry. study of composition/structure of matter, changes that matter undergoes, and forces that hold it together.

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CHEMISTRY SEMESTER 1 EXAM

REVIEW 1. Perform the following
metric to metric conversions. A. 32.6

cm = _____ m B. 0.0086 Kg = _____ g

2. Express these numbers in scientific

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notation or standard notation A. 0.000
000 000 154 m = B. 17 650 000g = 3.
What is the density of a piece of wood
whose mass is 35.0g and the volume
is 30.0 cm³?

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SEMESTER 1 EXAM REVIEW 1 ...

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Concepts: Terms in this set (181)

Chemistry. the study of the

composition, structure, and properties

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of matter, the processes that matter undergoes and the energy changes that accompany these processes.

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Terms in this set (83) How is the mole used to count particles of a substance? Avogadro's number is a proportion that relates molar mass on an atomic scale to physical mass on a

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Answers scale. Avogadro's number is ...

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Exam Preparations During Class. Day
1 (December 19th) review content on

Page 18/40

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REVIEW 1. A measure of the quantity
of matter is MASS 2.

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chemistry semester 1 exam review.
STUDY. PLAY. what are the steps to
the scientific method. 1)observation
2)question 3)hypothesis 4)experiment
5)results 6)conclusion. convert the

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following into scientific notation a)1500
b)1.52. a) 1.5×10^3 b) 1.52×10 . convert
into decimal notation a) 4.59×10^3
b) 5×10^2 . a)4590 b)500.

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Site for AP Chemistry students

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Chemistry - Mid Term Exam Review
Sheet #1 The midterm exam covers
chapters I - 4 & 9 - 11. You should read

Page 24/40

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through each chapter, look over old tests you still have. answer the following questions and do the calculations in order prepare yourself for the mid-term. 1. Define the following terms and describe where each is located. proton - neutron -

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Sheet #1

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this quiz on Quizizz. calcium
phosphate. Chemistry Semester 1

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Chemistry Semester Exam Review
MATTER Physical vs. Chemical
Properties and Changes 4.(A)
differentiate between physical and
chemical changes and properties; R 4

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Answers on the final 4.(B) identify extensive properties such as mass and volume & intensive properties such as density and melting point; 1 question on the final 1) What is a physical change? 2) What is a chemical change?

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semester exam. Terms in this set
(108). chemistry. study of

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Answers/structure of matter,
changes that matter undergoes, and
forces that hold it together.

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1st Semester Units: Unit 1:

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Chapters 1-3 Packet Chs 1 and 2
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Unit 2: Solutions Assignment Sheet
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Semester 1 - AP Chemistry
2016 PISD Chemistry Fall Semester
Exam Review KEY Page 1 of 8 1
Describe the differences between
solids, liquids, and gases. Draw a

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diagram of atoms in the solid state changing to liquid state changing to gaseous state. Solids have tightly packed particles, with high density.

2016 PISD Chemistry Fall Semester
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STRATEGY: Start by reading through your notes to refresh your memory on these topics. Then, use this review sheet as a starting point to identify the areas on which you need to spend more study time. For those areas, go

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Organic Chemistry I: Reactions and

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Overview. Organic I Review Packet.
Interactive Organic Reactions. Very
Extensive Material from Org I. General
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Summary of first semester reactions.
Summary of reactions for CHM 26100.
Reactions of different functional

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Answers (looks like Orgo threw up)

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