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is now becoming more and more required to investigate electronic structures of various solid materials in the bulk, on surfaces as well as at buried interfaces. The energy resolution was much improved in the last decade down to 1 meV in the low photon energy region.

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We report a joint negative ion photoelectron spectroscopy (NIPES) and computational study on the electronic structures and noncovalent interactions of a series of cyclodextrin-

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closo-dodecaborate dianion complexes, $\chi\text{-CD}\cdot\text{B}_{12}\text{X}_{12}^{2-}$ ($\chi = \alpha, \beta, \gamma$; $\text{X} = \text{H}, \text{F}$). The measured vertical/adiabatic detachment energies (VDE

Photoelectron spectroscopy and computational ...

Photoelectron spectroscopy (PES) is an experimental technique used to determine the relative energies of electrons in atoms and molecules. Photoelectron spectrometers work by ionizing samples using high-energy radiation (such as UV or x-rays) and then measuring the kinetic energies (

Photoelectron spectroscopy

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Photoelectron spectroscopy (PES) is the energy measurements of photoelectrons emitted from solids, gases, or liquids by the photoelectric effect. Depending on the source of ionization energy, PES can be divided accordingly into Ultraviolet Photoelectron Spectroscopy (UPS) and X-ray Photoelectron Spectroscopy (XPS). The source of radiation for UPS is a noble gas discharge lamp, usually a He discharge lamp.

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photoelectron spectroscopy (UPS) and X-ray photoelectron spectroscopy (XPS). XPS is also known under its former name of electron spectroscopy for chemical analysis (ESCA). UPS focuses on ionization of valence electrons while XPS is able to go a step further and ionize core electrons and pry them away.

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Abstract. Modern optoelectronics technologies are built on the basis of fundamental knowledge about electronic properties of semiconductor materials,

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which can be attained through accurate analyses made on single crystals of the materials as standard samples. In the cases of organic semiconductor materials, although technical difficulties have obstructed direct measurements of electronic properties in the organic semiconductor single crystals by photoelectron spectroscopy, recent advances ...

Photoelectron spectroscopy on single crystals of organic ...

Concepts illustrated: vibrational structure and Franck-Condon principle;

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adiabatic and vertical ionization energies; Koopmans's theorem; link between photoelectron spectra and molecular orbital diagrams; Morse potentials.. Carbon monoxide was one of the first molecules studied by ultraviolet photoelectron spectroscopy [1]. A typical HeI spectrum is shown in Figure 13.1.

Ultraviolet photoelectron spectrum of CO (Chapter 13

...

Electronic and photoelectron spectroscopy provides extraordinarily detailed information on the properties of molecules.

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This book presents the basic principles and experimental techniques employed in electronic and photoelectron spectroscopy.

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In the analytical technique of photoelectron spectroscopy (PES), a sample is ionized using high-energy radiation, and the kinetic energies of the ejected electrons (called photoelectrons) are measured. From this, we can

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determine the binding or ionization energy of each electron in the atom or molecule.

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Photoelectron spectroscopy is based on Einstein's photoelectric effect. 1 A photon can ionize an electron from a molecule if the photon has an energy greater than the energy holding the electron in the molecule. Any photon energy in excess of that needed for ionization is carried by the outgoing electron in the form of kinetic energy.

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