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© www.CHEMSHEETS.co.uk 12-June-2016 Chemsheets A2 1070 Page 4 TASK 4 – Predicting <sup>1</sup>H NMR spectra Compound Structure Number of signals Relative intensity of signals

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SECTION 1 – <sup>1</sup>H NMR Why compounds absorb radiowaves (background information beyond specifications) NMR (nuclear magnetic resonance) is a very powerful tool for identifying compounds. H, The nucleus of some atoms has nuclear spin (e.g. <sup>13</sup>C, <sup>19</sup>F, <sup>31</sup>P), although many atoms do not have any nuclear spin (e.g. <sup>12</sup>C).

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Propene reacts with HBr to form H. H reacts with sodium hydroxide to form I, and I reacts with warm acidified potassium dichromate (VI) to form J. The infra-red spectra of H, I and J are given below, but it does indicate which is - which. Identify the three compounds H, I and J, using the infra-red spectra below, and decide which spectrum belongs

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gizmo answers, of applied mechanics by r s khurmi, carrier aquasnap 30rba chiller manual, the ... SECTION 1 <sup>1</sup>H NMR - Weebly 1 Acid = H<sub>2</sub>O, base = NH<sub>3</sub> 2 Acid = HCl, base = H<sub>2</sub>O 3 Acid = HCOOH, base = KOH 4 Acid = HCl, base = CH<sub>3</sub>COOH 5 Acid = HCl, base = NH<sub>3</sub> 6 Acid = HCO<sub>3</sub><sup>-</sup>, base = OH<sup>-</sup> 7 Acid = H<sup>+</sup>, base = HCO<sub>3</sub><sup>-</sup> 8 Acid = H