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counting atoms related terms. Terms in this set (28) law of conservation of mass. states that mass is neither created nor destroyed during ordinary reactions or physical changes.

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Chapter 3 - Chemistry

Modern Chemistry Chapter 3. Atoms: The Building Blocks of Matter. law of conservation of mass- mass is neither created nor destroyed during ordinary chemical reactions or physical changes. e.g. $20 \text{ g A} + 20 \text{ g B} \rightarrow 40 \text{ g AB}$.

Modern Chemistry Chapter 3 Atoms: The

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Building Blocks of ...

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CHAPTER 3 REVIEW Atoms: The Building Blocks of Matter SECTION 2 SHORT ANSWER Answer the following questions in the space provided. 1. In cathode-ray tubes, the cathode ray is

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emitted from the negative electrode, which is called the cathode . 2. The smallest unit of an element that can exist either alone or in molecules containing the

3 Atoms: The Building Blocks of Matter

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chemical reactions or physical changes e.g.
20 g A + 20 g B 40 g AB law of definite
proportions- a chemical compound contains the
same elements in exactly the same proportions
by mass regardless of the size of the sample
or the source of the compound e.g. If 10
grams of A combine with 20 grams of B to form
compound AB, how ...

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Chapter 3 ...

CHAPTER 3 TEST Class Atoms: The Building
Blocks of Matter MULTIPLE CHOICE On the line
at the left of each statement, write the

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letter of the choice best completes the statement or answers the question. The behavior of cathode rays in a glass tube containing gas at low pressure led scientists to conclude that the rays were composed of a. energy

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