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Match description on the right to the correct crystal type on the left Modern chemistry chapter 10 review answers states of matter section 1. b ionic crystal (a) has mobile electrons in the crystal c covalent molecular crystal (b) is hard, brittle, and nonconducting a metallic crystal (c) typically has the lowest melting point of the four

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CHAPTER 10 REVIEW States of Matter SECTION 2 SHORT ANSWER Answer the following questions in the space provided. 1. a Liquids possess all the following properties except (a) relatively low density. (c) relative incompressibility. (b) the ability to diffuse. (d) the ability to change to a gas. 2. a.

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CHAPTER 5 REVIEW The Periodic Law SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. c In the modern periodic table, elements are ...

5 The Periodic Law

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Modern Chemistry 47 Chemical Bonding CHAPTER 6 REVIEW Chemical Bonding SECTION 4 SHORT ANSWER Answer the following questions in the space provided. 1. _____ In metals, the valence electrons are considered to be (a) attached to particular positive ions.

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