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Particles, Gamma

Rays, Positrons,

Electrons,

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Neutrons

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On Activity

Problem No 1 -

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**Reactions,**

**Radioactivity,**

**Fission and**

**Fusion Nuclear**

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Atomic Structure

In Just 14

Minutes!

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Understanding

how it works |

Physics Elearnin

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**Calculations**

half life

calculations 01

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~~Half-Life~~

~~Calculations:~~

~~Radioactive~~

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**The Race to  
Create a Star on  
Earth** STRUCTURE

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~~Expect the~~

~~unexpected~~

~~Steve Weatherall~~

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Atomic Structure  
and Properties!!*

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~~Part 3 : Law~~

~~Of Radioactive~~

~~Decay JEE/NEET~~

~~Organic~~ 22

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SECTION 4 SHORT

ANSWER Answer

the following

questions in the

space provided.

1. Match each of

the following

statements with

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the process(es)  
to which they  
apply, using one  
of the choices

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below: (1)

fission only (3)

both fission and  
fusion

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Answers

make augmented

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support for  
higher life.

This is not  
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Nuclear Decay

Anytime an  
unstable nucleus  
emits alpha or  
beta particles,  
the number of  
protons or  
neutrons  
changes. An  
example would be

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Chemistry (an  
Review  
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isotope of  
radium with the  
mass number  
226), which  
changes to  
radon-222 by  
emitting an  
alpha particle.

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Chapter 18 –

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Chemistry 289

Key Ideas

Answers 14.

Because protons

and neutrons

reside in the

nucleus of

atoms, they are

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called nucleons.

16. There are two forces among the particles within the nucleus. The first, called the electrostatic force, is the force between electrically charged particles. The

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Second force,  
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: August Dunbar.

nucleons.

protons and

neutrons.

nuclide. An atom

identified by

the number of

protons and

neutrons in its

nucleus. mass

defect. The

difference

between the mass

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Chemistry Figure

25-2 Both Pierre

and Marie Curie

played important

roles in

founding the

field of nuclear

chemistry. Marie

Curie went on to

show that unlike

chemical

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radioactivity is  
not affected by  
changes in

physical

conditions such  
as temperature  
and pressure.

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Chemistry

Section 1 Review

Answers the term

half-life, and

explain how it

relates to the

stability of a

nucleus. Define

and relate the

terms decay

series, parent

nuclide, and

daughter

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