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counting the s-electrons

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outer electrons).
Misunderstanding electron
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Confusing the terms
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level'. Dot-and-cross
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the two atoms bonded together and the shared pair of electrons. This example clearly shows the refinement of the GCSE ideas required at AS level. Clearly stating that the bond is the electrostatic attraction between nuclei and a shared pair of electrons, this example scored 2. 4 GCE Chemistry 8CH0 01

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Instructions • • ... 1 These
pieces of apparatus are used
in chemistry experiments.

... Draw a diagram to show
the arrangement of electrons
in an atom of lithium with
atomic number 3 and mass
number 7. (1) (e) A sample
of lithium contains 92.5% of
atoms with mass number 7 and

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minutes WCH01/01 Candidates
may use a calculator.

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| Question number | Answer | Mark |
|-----------------|--|--|
| 4 | The only correct answer is D (3p subshell 6, third quantum shell 18) | A is incorrect because 2 is the number of electrons in a 3p orbital and the 3d electrons have been omitted from the third quantum shell B is incorrect because 2 is the number of electrons in a 3p orbital C is incorrect because the 3d electrons have been omitted from the third quantum shell |

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and neutrons, surrounded by electrons, orbiting in shells 1.10 recall the relative mass and relative charge of a proton, neutron and electron 1.11 understand the terms atomic number, mass number, isotopes and relative

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