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Hi VCE Chemmers, This quick video explains
how SAC 3 - EEI on Aspirin, will work. Check
Edmodo for the rubric re: poster design on
Aspirin synthesis. I wish you all the best!
Mr J.

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yield doesn't really say much about purity, i guess the best way is to measure the melting point, and if its considerably below the range (i think ~138 is the expected of pure aspirin) then its impure (mainly due to the presence of water)

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